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Reviews

This publication is definitely not simple to begin on studying but quite fun to see. It really is full of knowledge and wisdom I am just effortlessly can get a satisfaction of studying a created pdf.
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paperback. Condition: New. Ship out in 2 business day, And Fast shipping, Free Tracking number will be provided after the shipment. Pages Number: 188 Publisher: Chemical Pub. Date :2008-03-01 version 2 2008-03-01 2nd printing. This book is written in Tianjin University Wang Zhenglie Advanced Engineering Specialist school with the book Physical Chemistry supporting experimental materials. This book has the introduction. experimental and appendix. The experimental part consists of 12 experiments involving chemical thermodynamics. chemical kinetics. electrochemistry and colloid chemistry and so on. Each experiment was divided into experimental purposes. the experimental principle. instruments and medicines. laboratory procedures. data processing and requirements. To enhance understanding of the content of the experiment. each experiment also contained the description of reflection and discussion questions. in order to achieve the purpose of broadening their horizons. Contents: Introduction I. The purpose of physical and chemical experiments. requirements and considerations Second. the physical and chemical analysis of three experimental error. the representation of the experimental data experimental determination of a Meyer molar mass of volatile substances in the regulation of experimental two baths and viscosity Experiment 3 Determination of the determination of liquid saturated vapor pressure enthalpy Experiment 4 Experiment 5 Determination of partial molar volume of binary solution in the determination of six experimental determination of chemical equilibrium constants of seven experimental method to measure the freezing point lower molar mass liquid-phase reaction equilibrium constant experiment eight Determination of experimental ninety-two yuan fully miscible liquid distillation curve experiment twelve yuan condensed phase diagram of the system measurement experiment eleven weak electrolyte conductivity method the dissociation constant of the experiment 12 of the original electromotive force measurement experiment thirteen catalytic decomposition of hydrogen peroxide Kinetics fourteen ethyl acetate saponification reaction rate constant and activation energy measurement experiment fifteen acetone iodination reaction rate constant and activation energy measurement...

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